

Next-generation catalysts for sustainable biodiesel production: a comprehensive review

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Abstract

Biodiesel, derived from renewable biomass, has emerged as a promising solution, particularly microalgae-based biodiesel, which offers high lipid productivity, CO₂ sequestration, and the ability to grow in non-arable land and wastewater. However, the commercialization of biodiesel faces challenges, including high production costs, energy-intensive processes, and scalability issues. This review explores the role of catalysts in biodiesel production, focusing on conventional homogeneous and heterogeneous catalysts, as well as next-generation catalysts such as nanocatalysts, enzymatic catalysts, and bifunctional catalysts. These advanced catalysts offer higher efficiency, reusability, and reduced environmental impact, aligning with Sustainable Development Goals (SDGs) 7 (Affordable and Clean Energy), SDG 13 (Climate Action), and SDG 12 (Responsible Consumption and Production). The integration of microalgae cultivation with biorefineries and waste streams, such as wastewater and flue gas, enhances sustainability by reducing costs and minimizing environmental impact. By addressing technical, economic, and environmental barriers, microalgae-based biodiesel can play a significant role in the global transition to a sustainable and low-carbon energy future, contributing to the achievement of multiple SDGs. This review highlights the potential of next-generation catalysts to revolutionize biodiesel production and underscores the importance of interdisciplinary research and policy frameworks in realizing this potential.

Keywords: biodiesel, sustainable, biomass, bioenergy, next generation catalyst

INTRODUCTION

The world's energy needs are expected to surge by nearly 50% by 2050, driven by increasing population, urbanization, and industrial activities [1, 2]. Fossil fuels, which currently supply more than 80% of global energy, are not only limited in availability but also major sources of greenhouse gas (GHG) emissions, exacerbating climate change [3]. The transportation sector alone contributes around 24% of global CO₂ emissions, underscoring the pressing need for sustainable energy solutions [4]. In this context, renewable energy sources like solar, wind, and biofuels have emerged as key alternatives to reduce reliance on fossil fuels. Among these, biodiesel—a renewable fuel produced from biomass—has gained significant traction due to its ability to integrate seamlessly into existing fuel infrastructure [5, 6]. This shift aligns with Sustainable Development Goal (SDG) 7 (Affordable and Clean Energy) and SDG 13 (Climate Action), which emphasize the adoption of clean energy and the reduction of carbon emissions to combat global warming.

Biodiesel is typically produced through the transesterification of triglycerides, which are found in vegetable oils, animal fats, and microbial oils, with an alcohol (usually methanol or ethanol) in the presence of a catalyst [7]. The choice of catalyst plays a critical role in determining the efficiency, cost, and environmental impact of the biodiesel production process. Conventional catalysts, such as homogeneous acids and bases, have been widely used due to their high reactivity and low cost [8]. However, these catalysts have several limitations, including non-reusability, generation of

wastewater, and environmental concerns, which have prompted the search for more sustainable and efficient alternatives [9].

Catalysts are essential for the transesterification reaction, as they lower the activation energy and increase the reaction rate, enabling the conversion of triglycerides into fatty acid methyl esters (FAMEs), the primary component of biodiesel [10]. Homogeneous catalysts, such as sulfuric acid and sodium hydroxide, are highly effective but suffer from several drawbacks, including difficulty in separation, corrosion of equipment, and generation of toxic waste [11]. Heterogeneous catalysts, on the other hand, offer advantages such as reusability, ease of separation, and reduced environmental impact, making them a more sustainable option [12].

In recent years, next-generation catalysts, including advanced heterogeneous catalysts, nanocatalysts, and enzymatic catalysts, have emerged as promising alternatives to conventional catalysts. These catalysts offer several advantages, such as higher activity, selectivity, and stability, as well as reduced environmental impact [13]. For example, nanocatalysts, which are characterized by their high surface area and unique electronic properties, have shown exceptional performance in biodiesel production, achieving high yields under mild reaction conditions [14]. Similarly, enzymatic catalysts, particularly immobilized lipases, offer the advantages of mild reaction conditions, high specificity, and reusability, making them an attractive option for sustainable biodiesel production [15].

Conventional Catalysts in Biodiesel Production

Homogeneous catalysts

Homogeneous catalysts, particularly acid and base catalysts, have been widely used in biodiesel production due to their high reactivity and low cost. Acid catalysts, such as sulfuric acid (H_2SO_4) and hydrochloric acid (HCl), are effective in catalyzing the transesterification of triglycerides with high free fatty acid (FFA) content, such as waste cooking oil and animal fats [11]. These catalysts are also capable of simultaneously catalyzing esterification and transesterification, making them suitable for low-quality feedstocks [9]. However, acid-catalyzed reactions are relatively slow and require high temperatures and long reaction times, which increase energy consumption and production costs [7]. Base catalysts, such as sodium hydroxide (NaOH) and potassium hydroxide (KOH), are more commonly used due to their higher reaction rates and milder reaction conditions compared to acid catalysts [10]. These catalysts are highly effective in converting refined vegetable oils into biodiesel, achieving conversion rates of over 98% under optimal conditions [8]. However, base catalysts are sensitive to the presence of FFAs and water, which can lead to soap formation and reduce biodiesel yield [16]. Additionally, homogeneous catalysts are difficult to separate from the reaction mixture, requiring extensive washing and purification steps that generate large amounts of wastewater [17].

Heterogeneous catalysts

Heterogeneous catalysts, which are typically solid materials, offer several advantages over homogeneous catalysts, including ease of separation, reusability, and reduced environmental impact [12]. Solid acid catalysts, such as sulfated zirconia and Amberlyst-15, are effective in catalyzing both esterification and transesterification reactions, making them suitable for low-quality feedstocks [11]. These catalysts are also more tolerant to FFAs and water, reducing the risk of soap formation and improving biodiesel yield [13].

Solid base catalysts, such as calcium oxide (CaO) and magnesium oxide (MgO), are widely used due to their high activity, low cost, and availability [18]. These catalysts are particularly effective in converting refined vegetable oils into biodiesel, achieving high yields under mild reaction conditions [12]. However, solid base catalysts are susceptible to leaching and deactivation in the presence of FFAs and water, which can limit their reusability and lifespan [19]. Additionally, the synthesis of heterogeneous catalysts often involves high-temperature calcination, which can be energy-intensive and environmentally unfriendly [13].

Enzymatic catalysts

Lipases, a type of biocatalyst, have gained considerable attention as a sustainable alternative to chemical catalysts due to their high specificity, mild reaction conditions, and ability to catalyze both esterification and transesterification reactions [15]. They are capable of converting a wide range of feedstocks—including low-quality oils and fats—into biodiesel with high yields and minimal byproducts [20]. Furthermore, enzymatic catalysts are biodegradable and non-toxic, making them environmentally friendly [21].

However, the use of free lipases presents certain limitations, such as high cost and low stability [15]. To address these challenges, researchers have explored various enzyme immobilization techniques, including adsorption, cross-linking, and encapsulation [20]. Immobilized lipases have demonstrated promising performance, producing high biodiesel yields under mild reaction conditions and showing excellent reusability over multiple cycles [21]. For instance, lipases immobilized on magnetic nanoparticles have exhibited high activity and stability in the transesterification of vegetable oils, achieving biodiesel yields of over 90% even after 10 cycles [14]. Despite these advancements, the high cost of enzyme production and immobilization remains a major barrier to the large-scale commercialization of enzymatic biodiesel production [15].

Next-Generation Catalysts for Biodiesel Production

Advanced heterogeneous catalysts

Advanced heterogeneous catalysts have emerged as a promising alternative to conventional catalysts due to their high activity, selectivity, and reusability. These catalysts are typically composed of mixed metal oxides, supported catalysts, and novel materials such as metal-organic frameworks (MOFs) and perovskites [13]. Mixed metal oxides, such as calcium-magnesium oxide (CaO-MgO) and zinc-aluminum oxide (ZnO-Al₂O₃), have shown excellent catalytic performance in biodiesel production, achieving high yields under mild reaction conditions [12]. The synergistic effect between the metal components enhances the catalytic activity and stability, making these materials suitable for industrial applications [18].

Supported catalysts, where the active phase is dispersed on a high-surface-area support such as silica, alumina, or carbon, offer several advantages, including improved dispersion of active sites, enhanced stability, and ease of separation [14]. For example, calcium oxide (CaO) supported on mesoporous silica has been shown to exhibit high activity and reusability in the transesterification of vegetable oils [19]. Similarly, MOFs, which are characterized by their high surface area, tunable porosity, and chemical functionality, have shown potential as catalysts for biodiesel production [17]. The unique properties of MOFs enable the design of catalysts with tailored active sites and improved mass transfer, leading to enhanced catalytic performance.

Nanocatalysts

Nanocatalysts, which are characterized by their high surface area, unique electronic properties, and quantum size effects, have gained significant attention in recent years due to their exceptional performance in biodiesel production [14]. Metal nanoparticles, such as gold (Au), silver (Ag), and palladium (Pd), have shown high activity and selectivity in the transesterification of triglycerides, achieving high biodiesel yields under mild reaction conditions [13]. The small size and high surface area of nanoparticles provide a large number of active sites, enhancing the catalytic activity and reducing the reaction time [12].

Carbon-based nanomaterials, such as carbon nanotubes (CNTs) and graphene, have also been explored as catalysts and catalyst supports for biodiesel production [17]. The high surface area, electrical conductivity, and chemical stability of these materials make them ideal for supporting metal nanoparticles and enhancing their catalytic performance [14, 22]. For example, palladium nanoparticles supported on graphene oxide have shown high activity and reusability in the

transesterification of vegetable oils, achieving biodiesel yields of over 95% under mild reaction conditions [13]. Despite their promising performance, the high cost and potential environmental risks of nanomaterials remain significant challenges for their large-scale application.

Bifunctional and hybrid catalysts

Bifunctional catalysts, which possess both acid and base sites, have emerged as a promising approach for simultaneous esterification and transesterification of low-quality feedstocks [13]. These catalysts are capable of converting FFAs and triglycerides into biodiesel in a single step, reducing the need for pre-treatment and improving the overall efficiency of the process [12]. For example, sulfated zirconia, which contains both acidic and basic sites, has shown high activity and selectivity in the conversion of waste cooking oil into biodiesel [11].

Hybrid catalysts, which combine the properties of chemical and enzymatic catalysts, have also been explored as a way to enhance the performance and sustainability of biodiesel production [20]. For example, lipase immobilized on a solid acid catalyst has been shown to exhibit high activity and stability in the transesterification of vegetable oils, achieving biodiesel yields of over 95% under mild reaction conditions [21]. The combination of chemical and enzymatic catalysis enables the efficient conversion of a wide range of feedstocks, including low-quality oils and fats, into biodiesel with minimal byproducts [15].

Sustainability and Environmental Impact of Next-Generation Catalysts

Green chemistry principles

The development of next-generation catalysts for biodiesel production aligns closely with the principles of green chemistry, which emphasize the design of chemical processes that minimize environmental impact and promote sustainability [23]. Green chemistry principles, such as the use of non-toxic and renewable materials, waste prevention, and energy efficiency, are critical for the development of sustainable catalysts [24]. For example, the use of biodegradable and non-toxic materials, such as lipases and carbon-based nanomaterials, reduces the environmental impact of biodiesel production [15].

The design of catalysts with high activity and selectivity also contributes to waste prevention by minimizing the formation of byproducts and reducing the need for purification steps [13]. Additionally, the use of renewable feedstocks, such as waste cooking oil and microalgae, further enhances the sustainability of biodiesel production by reducing reliance on fossil fuels and minimizing competition with food crops [25]. The integration of renewable energy sources, such as solar and wind energy, into the production process can also reduce the carbon footprint of biodiesel production and improve its overall sustainability [26].

Life Cycle Assessment (LCA)

Life cycle assessment (LCA) is a valuable tool for evaluating the environmental impact of next-generation catalysts and identifying opportunities for improvement [27]. LCA studies have shown that the use of advanced heterogeneous catalysts and nanocatalysts can significantly reduce the environmental impact of biodiesel production compared to conventional catalysts [13]. For example, the use of calcium oxide (CaO) as a solid base catalyst has been shown to reduce greenhouse gas emissions and energy consumption compared to homogeneous catalysts [12].

The environmental impact of enzymatic catalysts, particularly immobilized lipases, has also been evaluated using LCA. These studies have shown that enzymatic catalysts can reduce the carbon footprint of biodiesel production by minimizing energy consumption and waste generation [15]. However, the high cost of enzyme production and immobilization remains a significant barrier to the widespread adoption of enzymatic catalysts [20]. The development of low-cost and sustainable

enzyme production methods, such as the use of genetically engineered microorganisms, could further enhance the environmental benefits of enzymatic biodiesel production [21].

Economic viability

The economic viability of next-generation catalysts is a critical factor for their large-scale adoption in biodiesel production. The high cost of advanced catalysts, such as nanocatalysts and immobilized enzymes, remains a significant challenge for their commercialization [21]. However, the development of low-cost synthesis methods and the use of renewable and abundant materials, such as calcium oxide and carbon-based nanomaterials, can reduce the cost of catalyst production and improve their economic viability [12].

The integration of biodiesel production with waste streams, such as wastewater and flue gas, can also improve the economic viability of next-generation catalysts by reducing the cost of raw materials and providing additional revenue streams [28]. For example, the use of microalgae for biodiesel production can provide additional benefits, such as wastewater treatment and CO₂ sequestration, which can offset the cost of catalyst production and improve the overall economic feasibility of the process [29].

Policy supports and financial incentives, such as tax credits, subsidies, and grants, are also critical for promoting the adoption of next-generation catalysts and improving their economic viability [30]. For example, the European Union's Renewable Energy Directive (RED II) provides financial incentives for the production of advanced biofuels, including biodiesel produced using next-generation catalysts [31]. Similar policies and incentives in other regions could accelerate the development and deployment of sustainable catalysts for biodiesel production.

Challenges and Limitations

Technical challenges

Despite the significant advancements in next-generation catalysts for biodiesel production, several technical challenges remain. One of the primary issues is catalyst deactivation, which can occur due to fouling, poisoning, or sintering [13]. For example, heterogeneous catalysts such as calcium oxide (CaO) are prone to deactivation in the presence of free fatty acids (FFAs) and water, which can lead to the formation of soap and reduce catalytic activity [12]. Similarly, nanocatalysts, while highly active, are susceptible to aggregation and leaching, which can limit their reusability and lifespan [14]. Another technical challenge is the optimization of reaction conditions, such as temperature, pressure, and feedstock quality, to achieve high biodiesel yields and minimize byproduct formation [7]. For example, enzymatic catalysts, while highly specific, require precise control of reaction conditions to maintain their activity and stability [15]. Additionally, the scalability of next-generation catalysts from laboratory to industrial scale remains a significant challenge, as the performance of catalysts can vary significantly under different operating conditions [30].

Economic challenges

The high cost of next-generation catalysts is a major barrier to their widespread adoption in biodiesel production. Advanced catalysts, such as nanocatalysts and immobilized enzymes, are often expensive to produce and require complex synthesis methods [13]. For example, the synthesis of metal nanoparticles and carbon-based nanomaterials involves high-temperature and high-pressure processes, which can be energy-intensive and costly [14]. Similarly, the production and immobilization of lipases require specialized equipment and materials, which can increase the overall cost of enzymatic biodiesel production [20].

The economic viability of next-generation catalysts is further challenged by the low market price of biodiesel compared to fossil fuels [30]. In the absence of robust policy support and financial incentives, such as tax credits and subsidies, next-generation catalysts struggle to compete with

conventional catalysts and fossil fuels. Additionally, the high cost of raw materials, such as refined vegetable oils and waste cooking oil, can further increase the production cost of biodiesel and limit the economic feasibility of next-generation catalysts [17].

Environmental challenges

While next-generation catalysts offer several environmental benefits, such as reduced greenhouse gas emissions and waste generation, their production and use are not without environmental challenges. The synthesis of advanced catalysts, such as nanocatalysts and metal-organic frameworks (MOFs), often involves the use of toxic chemicals and high-energy processes, which can have a significant environmental impact [13]. For example, the production of metal nanoparticles can generate hazardous waste and contribute to environmental pollution if not properly managed [14, 32].

The disposal and recycling of spent catalysts is another environmental challenge. While next-generation catalysts are designed to be reusable, their eventual deactivation and disposal can pose environmental risks if not properly managed [12]. For example, the leaching of metal nanoparticles into the environment can have toxic effects on aquatic ecosystems and human health [13]. Additionally, the high energy consumption associated with the production and regeneration of catalysts can offset the environmental benefits of biodiesel production if the energy is derived from non-renewable sources [27].

Social and regulatory challenges

Public acceptance and awareness of next-generation catalysts and biodiesel production are critical for their successful commercialization. Despite their potential, next-generation catalysts and biodiesel are still relatively unknown to the general public, and misconceptions about their feasibility and benefits can hinder their adoption [33]. Educational campaigns and transparent communication about the advantages and limitations of next-generation catalysts are essential to build public trust and support.

Regulatory frameworks also play a crucial role in the development and deployment of next-generation catalysts. Inconsistent policies and lack of incentives for renewable energy in some regions create uncertainty for investors and hinder market growth [30]. For example, while the European Union has implemented ambitious targets for renewable energy adoption, other regions lag behind, creating an uneven playing field for next-generation catalysts and biodiesel producers [34]. Harmonizing policies and providing financial incentives, such as tax credits and subsidies, are essential to accelerate the commercialization of next-generation catalysts and biodiesel production.

Future Prospects and Research Directions

Emerging trends in catalyst development

The future of biodiesel production lies in the development of advanced catalysts that are not only highly efficient but also sustainable and cost-effective. One of the most promising trends is the use of machine learning (ML) and computational modeling to design and optimize catalysts. These technologies enable researchers to predict the performance of catalysts based on their structural and chemical properties, reducing the need for time-consuming and costly experimental trials [35]. For example, ML algorithms can analyze large datasets of catalytic reactions to identify optimal combinations of materials, reaction conditions, and catalyst configurations [36]. This approach has already been successfully applied to the design of heterogeneous catalysts for biodiesel production, achieving significant improvements in activity and selectivity [14].

Another emerging trend is the exploration of novel materials for catalyst synthesis. For instance, metal-organic frameworks (MOFs) and perovskites have shown great potential due to their high surface area, tunable porosity, and unique electronic properties [17]. MOFs, in particular, are highly versatile and can be functionalized with various active sites to enhance their catalytic performance.

For example, MOFs functionalized with sulfonic acid groups have demonstrated high activity in the esterification of free fatty acids (FFAs), making them suitable for low-quality feedstocks [12]. Similarly, perovskites, which are mixed metal oxides with a crystalline structure, have shown excellent stability and activity in transesterification reactions, particularly when doped with transition metals [13].

The development of bifunctional catalysts is another promising area of research. These catalysts possess both acid and base sites, enabling them to catalyze both esterification and transesterification reactions simultaneously [12]. For example, bifunctional catalysts based on mixed metal oxides, such as calcium-magnesium oxide (CaO-MgO), have shown high activity and selectivity in the conversion of waste cooking oil into biodiesel [18]. The integration of bifunctional catalysts into a single-step process can significantly reduce the complexity and cost of biodiesel production, making it more economically viable.

Integration with biorefineries

The concept of a biorefinery, where multiple value-added products are co-produced alongside biodiesel, is gaining traction as a way to improve the economic viability and sustainability of biodiesel production. In a biorefinery, the residual biomass after lipid extraction can be converted into biochar, biogas, or animal feed, while the extracted pigments and proteins can be used in the food, pharmaceutical, and cosmetic industries [37]. For example, microalgae-based biorefineries can produce high-value products such as astaxanthin, beta-carotene, and omega-3 fatty acids, which can generate additional revenue streams and offset the cost of biodiesel production [38].

The integration of wastewater treatment and CO₂ sequestration into microalgae cultivation further enhances the sustainability of the biorefinery concept. Microalgae can effectively remove nutrients such as nitrogen and phosphorus from wastewater, reducing the environmental impact of nutrient discharge [28]. Similarly, the use of flue gas as a carbon source for microalgae growth can mitigate CO₂ emissions from industrial processes, contributing to climate change mitigation [29]. These integrated approaches not only improve the economic feasibility of biodiesel production but also align with the principles of a circular economy, where waste is minimized, and resources are continuously reused.

Policy and market development

The successful commercialization of next-generation catalysts and biodiesel production requires robust policy support and market development. Governments play a pivotal role in promoting renewable energy adoption through policies such as renewable fuel standards (RFS), carbon pricing, and low-carbon fuel mandates [34]. For example, the European Union's Renewable Energy Directive (RED II) sets ambitious targets for renewable energy use in the transportation sector, creating a favorable market environment for biodiesel produced using next-generation catalysts [39]. Similar policies in other regions, such as the Renewable Fuel Standard (RFS) in the United States and the Biofuels Policy in India, have also contributed to the growth of the biodiesel industry [30].

Public-private partnerships and international collaborations are essential for accelerating the development and deployment of next-generation catalysts. For example, the Algae Biomass Organization (ABO) in the United States and the European Algae Biomass Association (EABA) are working to promote research, development, and commercialization of microalgae-based products [40]. These organizations provide a platform for collaboration between researchers, industry stakeholders, and policymakers, facilitating the exchange of knowledge and resources.

Raising public awareness about the benefits of biodiesel and next-generation catalysts is also critical for driving market demand. Educational campaigns and outreach programs can help dispel misconceptions and build consumer confidence in biodiesel as a sustainable and viable alternative to fossil fuels [41]. For example, highlighting the environmental benefits of biodiesel, such as reduced

greenhouse gas emissions and improved air quality, can encourage consumers to choose biodiesel over conventional diesel.

Research directions for overcoming challenges

Development of low-cost synthesis methods

The high cost of advanced catalysts, such as nanocatalysts and immobilized enzymes, remains a significant barrier to their commercialization. Research efforts should focus on developing low-cost synthesis methods using renewable and abundant materials, such as calcium oxide and carbon-based nanomaterials [12]. For example, the use of agricultural waste as a precursor for catalyst synthesis can reduce costs and improve sustainability [13].

Improvement of catalyst stability and reusability

Catalyst deactivation is a major challenge that limits the lifespan and reusability of next-generation catalysts. Research should focus on improving the stability of catalysts through surface modification, doping, and the use of protective coatings (Gardy *et al.*, 2019). For example, the use of graphene oxide as a support material can enhance the stability and reusability of metal nanoparticles [17].

Optimization of reaction conditions

The performance of next-generation catalysts is highly dependent on reaction conditions, such as temperature, pressure, and feedstock quality. Research should focus on optimizing these conditions to achieve high biodiesel yields and minimize byproduct formation [7]. For example, the use of microwave-assisted transesterification can reduce reaction times and energy consumption [42].

Exploration of alternative feedstocks

The high cost of refined vegetable oils and competition with food crops are significant challenges for biodiesel production. Research should focus on exploring alternative feedstocks, such as waste cooking oil, animal fats, and microbial oils, which are more sustainable and cost-effective [25]. For example, the use of genetically engineered microorganisms to produce lipids from lignocellulosic biomass can provide a renewable and abundant feedstock for biodiesel production [43].

Development of sustainable enzyme production methods

The high cost of enzyme production and immobilization remains a significant barrier to the commercialization of enzymatic biodiesel production. Research should focus on developing low-cost and sustainable enzyme production methods, such as the use of genetically engineered microorganisms and agricultural waste as substrates [20]. For example, the use of agro-industrial waste as a substrate for lipase production can reduce costs and improve sustainability [21].

CONCLUSIONS

The development of next-generation catalysts for sustainable biodiesel production represents a critical step toward addressing the global energy crisis and mitigating the environmental impact of fossil fuels. Over the past decade, significant advancements have been made in the design and synthesis of advanced catalysts, including heterogeneous catalysts, nanocatalysts, and enzymatic catalysts. These catalysts offer numerous advantages over conventional catalysts, such as higher activity, selectivity, and reusability, as well as reduced environmental impact. However, several challenges remain, including high production costs, catalyst deactivation, and scalability issues, which must be addressed to enable their widespread adoption in industrial biodiesel production.

The integration of next-generation catalysts with biorefineries and waste streams, such as wastewater and flue gas, offers a promising pathway to improve the economic viability and sustainability of biodiesel production. By co-producing high-value products and utilizing renewable feedstocks, biorefineries can reduce production costs and minimize waste, aligning with the principles of a circular economy. Additionally, the use of advanced technologies, such as machine learning and computational modeling, can accelerate the development of novel catalysts and optimize their performance, further enhancing the efficiency and sustainability of biodiesel production.

Policy support and public awareness are also critical for the successful commercialization of next-generation catalysts. Governments play a pivotal role in promoting renewable energy adoption through policies such as renewable fuel standards, carbon pricing, and low-carbon fuel mandates. Public-private partnerships and international collaborations can facilitate the exchange of knowledge and resources, driving innovation and market development. Educational campaigns and outreach programs can help build public trust and support for biodiesel as a sustainable and viable alternative to fossil fuels.

In conclusion, the future of biodiesel production lies in the development of sustainable and efficient catalysts that can overcome the technical, economic, and environmental challenges associated with conventional catalysts. By addressing these challenges and leveraging emerging technologies, next-generation catalysts have the potential to revolutionize the biodiesel industry and contribute to a more sustainable and low-carbon energy future. Collaborative efforts between researchers, industry stakeholders, and policymakers are essential to accelerate the development and deployment of these catalysts and realize their full potential.

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